

concl A1

35 U.S.C. § 371. The entire contents of each of the above-identified applications are hereby incorporated by reference.--

Please replace the paragraph beginning on page 2, line 5 through page 3, line 2, with the following rewritten paragraph:

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--For the purpose of overcoming these problems, non-azide type gas generants have been under development in place of sodium azide type. For example, JP-A 3-208878 discloses a composition which comprises a main component of oxygen-containing oxidizer such as tetrazole, triazole, or their metal salts combined with ammonium perchlorate or sodium nitrate and an auxiliary component of a metal oxide such as V_2O_5 , CuO or Fe_2O_3 . The metal oxide mentioned forms a solid combustion product, which is easily removed by filtration when undesired components are removed by filtration prior to release the formed gas into the bag in an air bag system, and converts CO which is toxic to human bodies generated from the nitrogen-containing organic compound to CO_2 . JP-B 64-6156 and JP-B 64-6157 disclose a gas generant which contains, as the main component, a metal salt of bitetrazole compound that does not contain hydrogen. JP-A-5-213687 further discloses a gas generant having a transition metal complex of aminoarazole as the main component. These non-azide type compounds disclosed in the series of prior art references are characteristic in that the concentration of carbon monoxide released is low since the carbon number in the one molecule is

small; however, the amount of nitrogen oxides, toxic to human bodies, increases in all the cases and the performance is not satisfactory in respect of the inflation period of the bag.--

Please replace the paragraph beginning on page 30, line 25 through page 31, line 3, with the following rewritten paragraph:

--Example 4-1 was repeated except that manganese dioxide and copper oxide were not added to evaluate the concentration of nitrogen oxides in the gas generated in the tank. The burning pressure and period were of desired values. The concentration of nitrogen oxides in the gas generated in the tank was not less than 2,000 ppm.--

Please replace the paragraph on page 32, lines 6-12, with the following rewritten paragraph:

--Example 4-1 was repeated except that the amount of catalyst addition was changed to 25 parts of manganese dioxide and 2 parts of copper (II) oxide to evaluate the concentration of nitrogen oxides in the gas generated in the tank. The burning pressure and period were of desired values. The concentration of nitrogen oxides in the gas generated in the tank was 630 ppm.--

IN THE CLAIMS:

Please add the following new claims:

--25. A gas generant composition comprising a nitrogen-

containing organic compound, an oxygen-containing inorganic oxidizer, and a third component of at least one selected from the following (1) through (4):

(1) manganese dioxide having a specific surface area not less than 50 m²/g;

(2) copper oxide having a specific surface area not less than 1 m²/g;

(3) a molybdenum compound of at least one selected from the group consisting of molybdenum dioxide, molybdenum trioxide, molybdic acid and ammonium molybdate; and

(4) a mixture of manganese dioxide and at least one metal oxide selected from the group consisting of copper oxides, cobalt oxides, iron oxides and silver oxides;

wherein the composition does not include an alkali metal salt or alkaline earth metal salt of hydrazoic acid.--

--26. The composition according to Claim 1, further comprising a decomposition promoter for accelerating the decomposition of the nitrogen-containing organic compound. --

--27. A gas generant composition according to claim 1, wherein the nitrogen containing organic group contains an amido group, and the oxygen containing inorganic oxidizer is an oxyhalogenate.--

--28. The composition according to Claim 1, wherein the nitrogen-containing organic compound is selected from the group consisting of urea, aminoguanidine bicarbonate, biuret, and hydrazides.--

--29. The composition according to Claim 1, wherein the third component is manganese dioxide having a specific surface area of 100 to 300 m²/g.--

--30. The composition according to Claim 1, wherein the third component is copper oxide having a specific surface area of 1 to 100 m²/g.--

--31. The composition according to Claim 1, wherein the third component is a molybdenum compound of at least one selected from the group consisting of molybdenum trioxide, molybdic acid and ammonium molybdate.--